

STUDY MODULE DESCRIPTION FORM		
Name of the module/subject Chemistry		Code 1011101331011000133
Field of study Engineering Management - Full-time studies -	Profile of study (general academic, practical) (brak)	Year /Semester 2 / 3
Elective path/specialty -	Subject offered in: Polish	Course (compulsory, elective) elective
Cycle of study: First-cycle studies	Form of study (full-time, part-time) full-time	
No. of hours Lecture: 30 Classes: 15 Laboratory: - Project/seminars: -		No. of credits 4
Status of the course in the study program (Basic, major, other) (brak)		(university-wide, from another field) (brak)
Education areas and fields of science and art		ECTS distribution (number and %)
Responsible for subject / lecturer: Prof dr hab. inż. Zenon Łukaszewski email: e-mail: office_chte@put.poznan.pl, tel. tel. (0*61) 6652 786, fax (0*61) 6652 571, Faculty of Chemical Technology ul. Piotrowo 3, 60-965 Poznań,		
Prerequisites in terms of knowledge, skills and social competencies:		
1	Knowledge	General chemistry on a high school level
2	Skills	Basic fluency in English language
3	Social competencies	Ability to work in a team
Assumptions and objectives of the course: Assumptions and objectives of the course: The aim of the course is to gain the knowledge from the area of chemical foundations of material science i.e. metal corrosion, synthetic polymers and lubricants		
Study outcomes and reference to the educational results for a field of study		
Knowledge: 1. Understanding of mechanism of metal corrosion and methods of corrosion prevention. Understanding of polymers structure and link between polymers structure and its properties. - [K04_Inz_AW02, K07_Inz_AW05]		
Skills: 1. Recognition of chemical formulas and language of chemical reactions - [K01_InzAU2, K01_InzAU7]		
Social competencies: 1. Ability to communicate in English language in the area of metal corrosion and polymers. Ability to communicate with chemists - [K01_InzAK01]		
Assessment methods of study outcomes		
Current assessment during classes.		
Course description		
Corrosion of metals. Electrochemical mechanism of corrosion. Anodic and cathodic reactions. Electrolyte. Protection of metals against corrosion. Coatings. Metallic coatings. Protectors. Cathodic protection. Anodic protection. Corrosion inhibitors. Chemical structure of polymers. Linear and cross-linked polymers. Termoplasticity of polymers. Chemical structures of popular polymers. Language of chemistry as an element of engineer knowledge.		

Basic bibliography:		
Additional bibliography:		
Result of average student's workload		
Activity	Time (working hours)	
1. Lecture	30	
2. Classes	15	
3. Consultations	10	
4. Preparation for classes	25	
5. Preparation for assessment of classes	6	
6. Preparation for assessment of lectures	10	
7. Final assessment of lectures	2	
8. Final assessment of classes	2	
Student's workload		
Source of workload	hours	ECTS
Total workload	100	4
Contact hours	59	2
Practical activities	15	1